

12. Atomic Nucleus, 4

13. Nuclear Interactions, 1

lecture 37, November 27, 2017

housekeeping

Coming attractions

This week:

chapter 12 homework due Wed 11/29

HW workshop tomorrow 11/28

End game:

I've made some adjustments to the schedule...stay tuned, now, week by week

exam #3 is Friday, December 1: chapters 7,8,9,12,13



today

Atomic nucleus - finishing

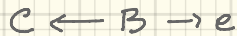
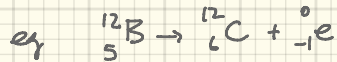
Nuclear Interactions - beginning



β decay.

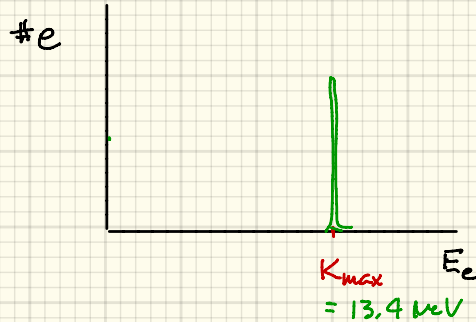
β decay confusing since 19-teens

measurements of $E(\beta)$ well established before Bohr model.

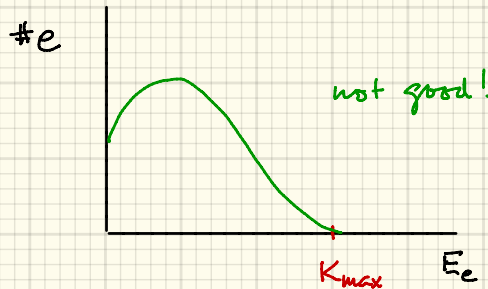


$$Q = [M({}_{5}^{12}\text{B}) - M({}_{6}^{12}\text{C})]c^2$$

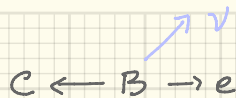
$$Q = 13.37 \text{ MeV}$$



Experiment(s):



Pauli's half-hearted idea



something else...
invisible

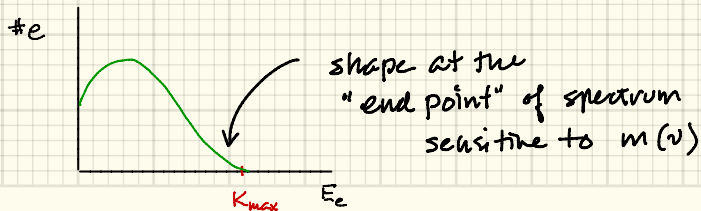
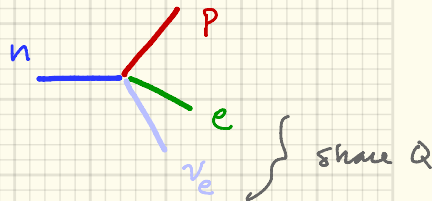
he called it the "neutron" -- nobody paid attention

↳ when Chadwick found our neutron
a new name was required.

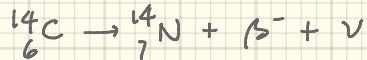
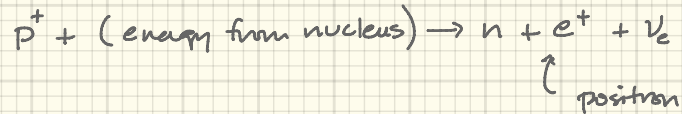
1934 Enrico Fermi -- wrote the complete theory of what he called the
"little neutron"... neutrino

inside the nucleus

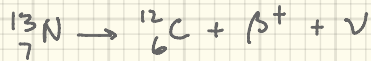
mass ν ? zero... we thought. not now. $\sim 10^3$ eV



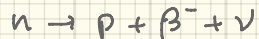
Inside nucleus... other β decays happen.



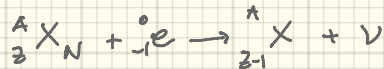
beta decay. → important for carbon-dating



positron decay



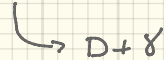
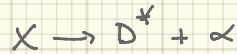
free neutron decay.



electron capture

Gamma Decay.

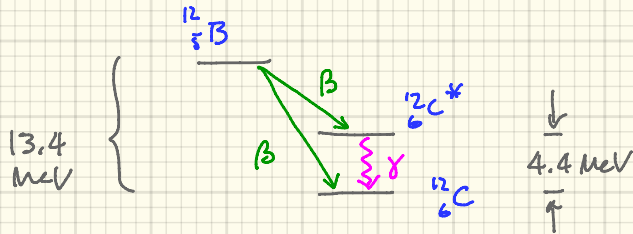
often:



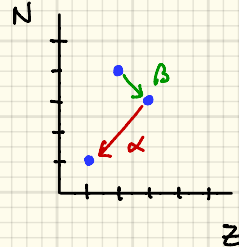
--- left in an excited state

and, collisions can excite a nucleus

Example ... just like atomic spectra.



cascades:



and so on

Radioactivity.

- natural → found in nature... stars
- artificial → lab-created, reactors, nuclear explosions

change A: α decay $A \rightarrow A-4$

change Z: α decay $Z \rightarrow Z-2$
 β decay $Z \rightarrow Z+1$

3 naturally occurring radioactive "chains"

${}_{92}^{238}\text{U}$ $T_{1/2} = 4.47 \times 10^9 \text{ y}$ → ${}_{82}^{206}\text{Pb}$ "uranium chain"

${}_{92}^{235}\text{U}$ $T_{1/2} = 7.04 \times 10^8 \text{ y}$ → ${}_{82}^{207}\text{Pb}$ "actinium chain"

${}_{90}^{232}\text{Th}$ $T_{1/2} = 1.41 \times 10^{10} \text{ y}$ → ${}_{82}^{208}\text{Pb}$ "thorium chain"

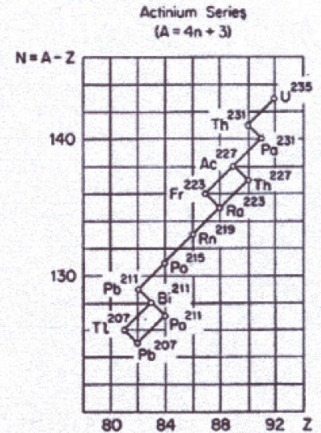
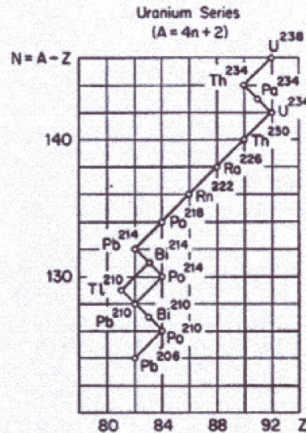
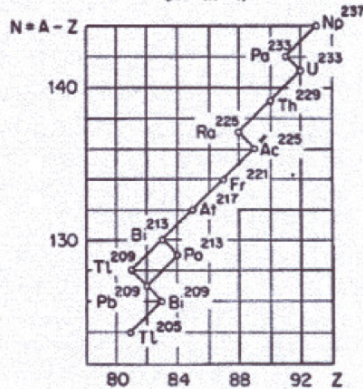
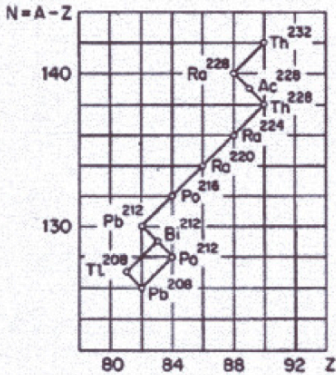
one artificial chain - "trans-uranic" Np

${}_{93}^{239}\text{Np}$ $T_{1/2} = 2.14 \times 10^6 \text{ y}$ → ${}_{83}^{209}\text{Bi}$ "neptunium chain"

Thorium Series
($A = 4n$)



Neptunium Series
($A = 4n + 1$)

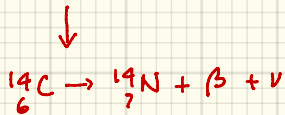


↑
not found in
nature -
artificially
produced.



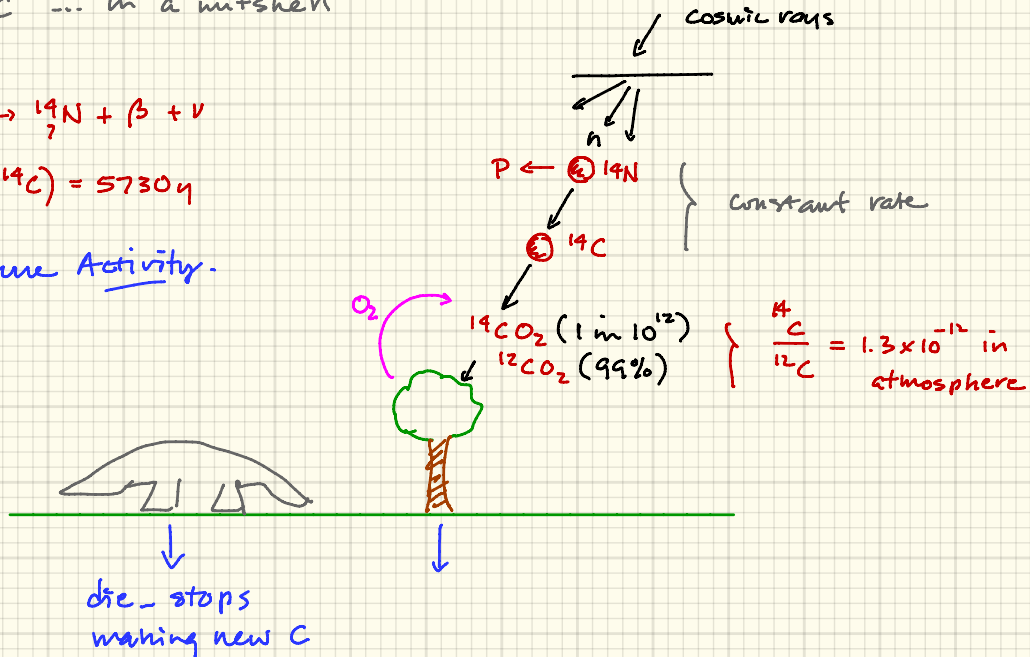
Radioactive Dating

^{14}C ... in a nutshell



$$T_{1/2}(^{14}\text{C}) = 5730 \text{ y}$$

measure Activity.



Example

Capture a sample of CO_2 from atmosphere into

$$V = 200 \text{ cm}^3$$

$$P = 2.0 \times 10^4 \text{ Pa} \sim 10^5 \text{ atm}$$

$$T = 295 \text{ K}$$

assume ^{14}C counting efficiency is perfect \rightarrow after 1 week, what is count?

$$n = \frac{PV}{RT} = \# \text{ moles} = \frac{(2 \times 10^4 \text{ N/m}^2)(2 \times 10^{-4} \text{ m}^3)}{(8.314 \text{ J/mol K})(295 \text{ K})} = 1.63 \times 10^{-3} \text{ mol}$$

$$N = N_A n = 9.82 \times 10^{20} \text{ molecules } \text{CO}_2 = \# \text{ atoms of } {}_6\text{C}$$

$$\# {}^{14}\text{C} = (1.3 \times 10^{-12})(9.82 \times 10^{20}) = 1.28 \times 10^9 \text{ atoms } {}_6^{14}\text{C}$$

$$R_0 = N_0 \lambda = (1.28 \times 10^9) \left(\frac{\ln 2}{T_{1/2}} \right) = (1.28 \times 10^9) \left(\frac{0.693}{(5730 \text{ y})} \right) \left(\frac{1 \text{ y}}{(\pi \times 10^7 \text{ s})} \right)$$

$$R_0 = 4.91 \times 10^{-3} \text{ decay/s}$$

$$R = 2969 \text{ decays/wh.}$$

Example

put piece of old wood into same device ...

after 1 wh, 1420 counts are recorded. how old is the wood?

$$R = R_0 e^{-\lambda t}$$

$$R_0 = 2969 \text{ s}^{-1} \quad \text{now}$$

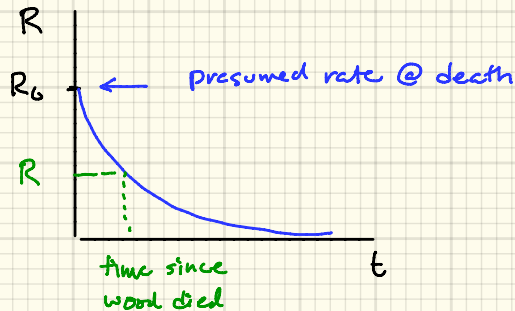
$$R = 1420 \text{ s}^{-1}$$

$$t = \frac{1}{\lambda} \ln \left(\frac{2969}{1420} \right)$$

$$T_{1/2} = \frac{\ln 2}{\lambda} = \frac{0.693}{\lambda}$$

$$t = \frac{5730}{0.693} \ln \left(\frac{2969}{1420} \right)$$

$$t = 6099 \text{ y}$$





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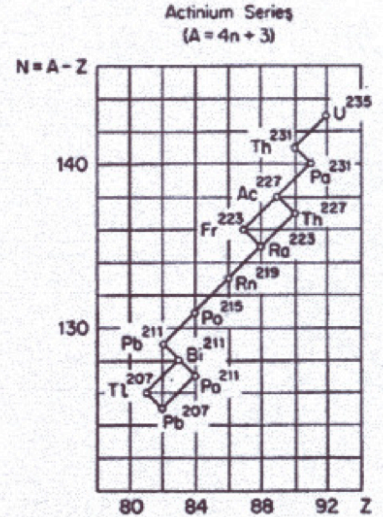
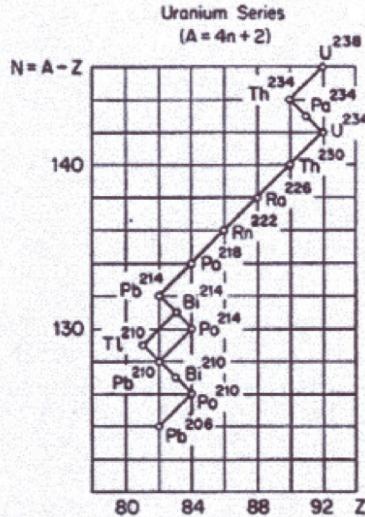
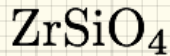
Quantity: 1

Add to List

Diamond jewelry at savings that shine



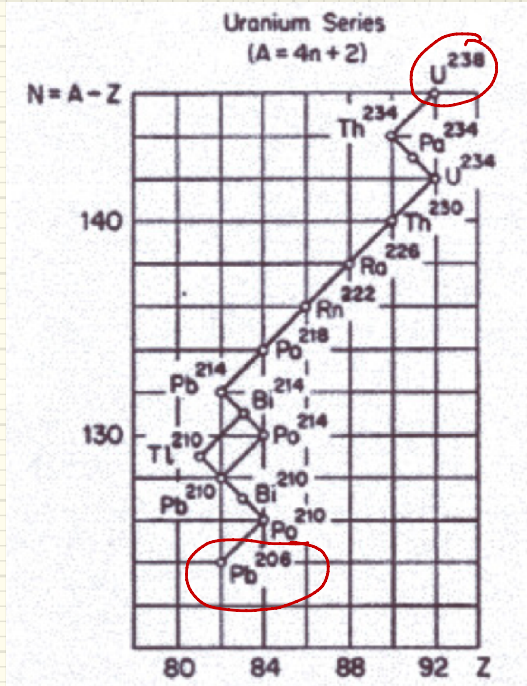
Zircon



1																	18
H	2											13	14	15	16	17	He
Li	Be											B	C	N	O	F	Ne
Na	Mg	3	4	5	6	7	8	9	10	11	12	Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Lr	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No

Uranium-Lead dating



Assume all ^{206}Pb in a sample comes from ^{238}U decay... *

$$\# \text{ } ^{238}\text{U} \text{ originally} - \# \text{ } ^{238}\text{U} \text{ today} = \# \text{ } ^{206}\text{Pb}$$

$$N_0 = \# \text{ } ^{238}\text{U} \text{ originally.}$$

$$N_0 e^{-\lambda t} = \# \text{ } ^{238}\text{U} \text{ today}$$

$$R = \frac{\# \text{ } ^{238}\text{U}}{\# \text{ } ^{206}\text{Pb}} = \frac{N_0 e^{-\lambda t}}{N_0 - N_0 e^{-\lambda t}}$$

solve for t

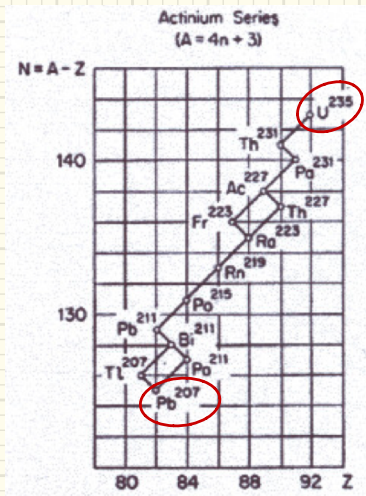
$$t = \frac{1}{\lambda} \ln \left(\frac{1}{R} + 1 \right) \quad \lambda = \frac{T_{1/2}}{0.693}$$

$$t = \frac{T_{1/2}}{0.693} \left(\frac{1}{R} + 1 \right) \quad T_{1/2} (^{238}\text{U}) = 4.5 \times 10^9 \text{ y}$$

* probably... There are other techniques

Pb dating, another way ^{204}Pb : stable

nothing decays to it
it doesn't decay. } abundance constant



But:

$$^{238}\text{U} \quad T_{1/2} = 4.47 \times 10^9 \text{ y} \longrightarrow ^{206}\text{Pb}$$

$$^{235}\text{U} \quad T_{1/2} = 0.7 \times 10^9 \text{ y} \longrightarrow ^{207}\text{Pb}$$

Earth is about $4.5 \times 10^9 \text{ y} \longrightarrow$

$6 \times T_{1/2} (^{235}\text{U}) \Rightarrow$ most is gone

^{238}U is still decaying from original abundance

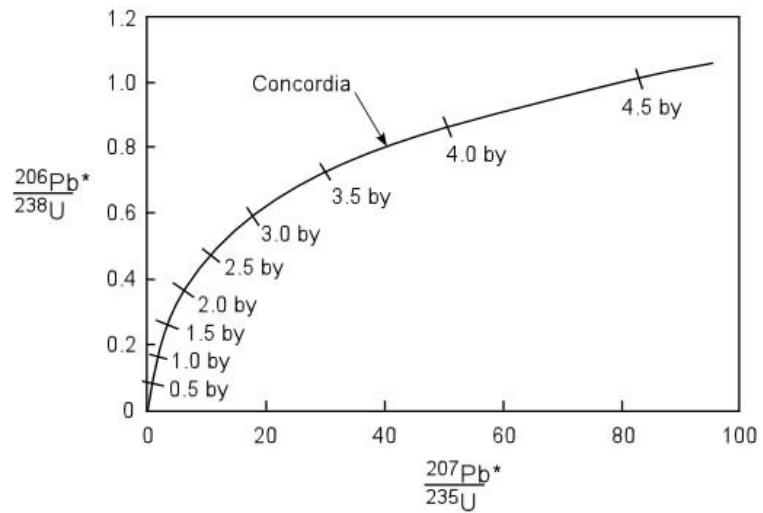
$$\frac{^{207}\text{Pb}}{^{204}\text{Pb}} = \text{constant} \quad \& \quad \frac{^{206}\text{Pb}}{^{204}\text{Pb}} \text{ still increasing}$$

compared with

abundance

- dates lead ore
- meteorites $\sim 4.5 \text{ By}$
- oldest rocks $\sim 4 \text{ By}$

Stik another way

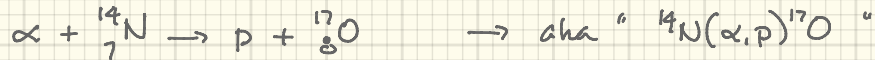


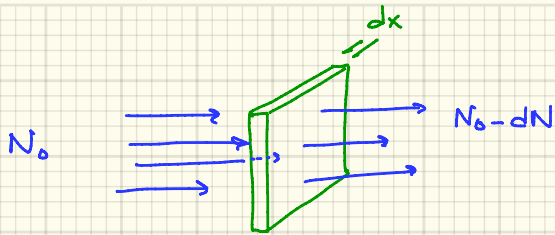
NUCLEAR REACTIONS

scattering particles from nuclei

- sometimes altering the trajectory
- sometimes completely changing the particles

Rutherford ... again ... proton discovery





total # target nuclei

nAx
 $\frac{\#}{\text{volume}} [m^{-3}]$ having cross section σ

total "area" presented to the beam

$$\sigma nAx$$

$$\frac{\text{collisions}}{\text{beam particles}} = \frac{\text{Rate of collisions } [s^{-1}]}{\text{Rate of beam } [s^{-1}]} = \frac{R}{R_0} = \frac{\sigma nAx}{A} = \sigma n x$$

$$-\frac{dN}{N} = \frac{nA\sigma dx}{A} = n\sigma dx$$

$$-\int_{N_0}^N \frac{dN}{N} = -n\sigma \int_0^x dx \Rightarrow \ln N/N_0 = -n\sigma x$$

$$N = N_0 e^{-n\sigma x}$$

remember $[\sigma] = \text{barn}$

Reaction kinematics ... low energy.



↑
at rest \Rightarrow "lab frame"

$$E_o = E_f$$

$$m_a c^2 + K_a + m_B c^2 + K_B = m_c c^2 + K_c + m_D c^2 + K_D$$

↑
= 0

rearrange mass energies = kinetic energies.

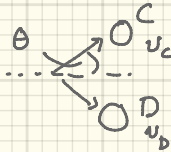
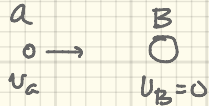
$$m_a c^2 + m_B c^2 - (m_c c^2 + m_D c^2) = K_c + K_D - K_a \equiv "Q"$$

$Q > 0 \Rightarrow$ final $K >$ initial K
(exothermic) \Rightarrow mass deficit

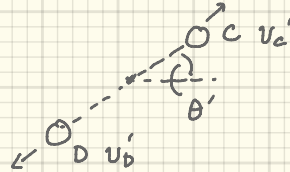
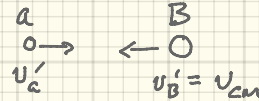
$Q < 0 \Rightarrow$ K converted into mass energy
(endothermic)

"threshold" required - Ka center of mass frame - cm - analysis

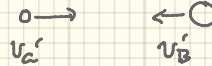
LAB



CM



transformation:



$$\vec{p}'_a = -\vec{p}'_B$$

$$m_a |v'_a| = m_B |v'_B|$$

$$v_B = v_{cm}$$

$$\left. \begin{array}{l} \xrightarrow{v_a} \\ \xrightarrow{v'_a} \quad \xleftarrow{v_B} \end{array} \right\} \begin{array}{l} v'_a = v_a - v_B \\ v'_a = v_a - v_{cm} \end{array} \longrightarrow$$

$$m_a |v_a'| = m_B |v_B'|$$

$$v_a' = v_a - v_{cm} \longrightarrow v_B' = \frac{m_a}{m_B} v_a' = \frac{m_a}{m_B} (v_a - v_{cm}) = v_{cm}$$

$$v_{cm} \left(1 + \frac{m_a}{m_B}\right) = \frac{m_a}{m_B} v_a' = v_{cm} \left(\frac{m_B + m_a}{m_B}\right)$$

$$\Rightarrow v_{cm} = \left(\frac{m_B}{m_B + m_a}\right) \frac{m_a}{m_B} v_a$$

$$v_{cm} = \frac{m_a}{m_a + m_B} v_a$$

In cm frame:

$$E_0 = E_f'$$

$$m_a c^2 + \frac{1}{2} m_a v_a'^2 + m_B c^2 + \frac{1}{2} m_B v_{cm}^2 = m_C c^2 + \frac{1}{2} m_C v_C'^2 + m_D c^2 + \frac{1}{2} m_D v_D'^2$$

$$\frac{1}{2} m_a (v_a - v_{cm})^2 + m_a c^2 + m_B c^2 + \frac{1}{2} m_B v_{cm}^2 = m_C c^2 + \frac{1}{2} m_C v_C'^2 + m_D c^2 + \frac{1}{2} m_D v_D'^2$$

"Threshold" \equiv C & D are produced with no kinetic energies $v_C' = v_D' = 0$

$$\frac{1}{2} m_a \left(v_a - \frac{m_a}{m_a + m_B} v_a\right)^2 + \frac{1}{2} m_B \left(\frac{m_a}{m_a + m_B} v_a\right)^2 = m_C c^2 + m_D c^2 - m_a c^2 - m_B c^2 = -Q$$

$$\frac{1}{2} m_a \left(v_a - \frac{m_a}{m_a + m_B} v_a \right)^2 + \frac{1}{2} m_B \left(\frac{m_a}{m_a + m_B} v_a \right)^2 = m_C \dot{c}^2 + m_D \dot{c}^2 - m_a \dot{c}^2 - m_B \dot{c}^2 = -Q$$

$$\frac{1}{2} m_a \left(v_a - \frac{m_a v_a}{m_a + m_B} \right)^2 + \frac{1}{2} m_B \frac{m_a^2 v_a^2}{(m_a + m_B)^2} = -Q$$

$$\frac{1}{2} m_a \left[v_a^2 + \frac{m_a^2 v_a^2}{\Delta^2} - 2 v_a^2 \frac{m_a}{\Delta} \right] + \frac{1}{2} \dots$$

$$\frac{1}{2} v_a^2 \left[m_a + \frac{m_a^3}{\Delta^2} - 2 \frac{m_a^2}{\Delta} + \frac{m_B m_a^2}{\Delta^2} \right]$$

$$\frac{1}{2} \frac{v_a^2}{\Delta^2} \left[m_a \Delta^2 + m_a^3 - 2 m_a^2 \Delta + m_B m_a^2 \right]$$

$$\frac{1}{2} \frac{v_a^2 m_a}{\Delta^2} \left[(m_a + m_B)^2 + m_a^2 - 2 m_a (m_a + m_B) + m_B m_a \right]$$

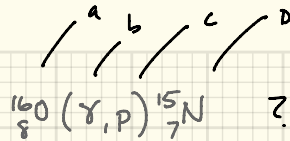
$$\frac{1}{2} \frac{v_a^2 m_a}{\Delta^2} \left[\cancel{m_a^2} + m_B^2 + 2 \cancel{m_a m_B} + \cancel{m_a^2} - 2 \cancel{m_a^2} - 2 \cancel{m_a m_B} + m_B m_a \right]$$

$$\frac{1}{2} \frac{v_a^2}{\Delta} m_a m_B \left[m_B + m_a \right] = \frac{1}{2} v_a^2 \frac{m_a m_B}{m_a + m_B} = -Q$$

$$\Rightarrow -Q \left(\frac{m_a + m_B}{m_B} \right) = \frac{1}{2} m_a v_a^2 = K_a^{\text{threshold}}$$

example:

Q fr



$$Q = [m_a + m_b - (m_c + m_d)] c^2$$

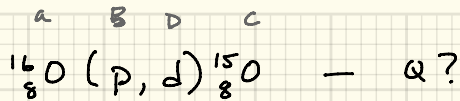
$$= [15.994915 \text{ u} - 0 \text{ u} - (15.000108 \text{ u} + 1.007825 \text{ u})] c^2 \quad 931.5 \frac{\text{MeV}}{c^2 \text{ u}}$$

$$Q = -12.13 \text{ MeV}$$

Threshold γ kinetic energy?

$$K_{\gamma}^{\text{th}} = -Q \left(\frac{m_c + m_d}{m_B} \right)$$

$$K_{\gamma}^{\text{th}} = (12.13) \left(\frac{15.994915 + 0}{15.99415} \right) \text{ u} = 12.13 \text{ MeV.}$$



$$Q = [m_a + m_B - (m_c + m_D)] c^2$$

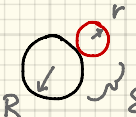
$$= [15.994915 \text{ u} + 1.007825 \text{ u} - 15.003070 \text{ u} - 2.014102 \text{ u}] c^2 \frac{931.5 \text{ MeV}}{c^2 \text{ u}}$$

$$Q = -13.44 \text{ MeV}$$

$$K_p^{\text{th}} \approx -Q \left(\frac{16\text{u} + 1\text{u}}{16\text{u}} \right) = 13.44 \text{ MeV} \left(\frac{17}{16} \right) = 14.26 \text{ MeV.}$$

There's actually more: proton must overcome electrostatic repulsion

estimate...



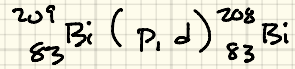
$$S = R + r \quad R = r_0 A^{1/3}$$

$$S \approx r_0 (A^{1/3} + 1) \quad r_0 = 1.4 \text{ fm.}$$

$$E_c = \frac{1}{4\pi\epsilon_0} \frac{Ze \cdot e}{S} = \frac{1.44 \text{ MeV} \cdot \text{fm}}{1.4 \text{ fm}} \frac{Z}{A^{1/3} + 1}$$

$${}^{16}_8\text{O} (p, d) {}^{15}_8\text{O}: \quad E_c = 2.34 \text{ MeV} \quad \text{so} \quad K_p^{\text{th}}? > E_c \quad \checkmark \quad \text{will happen!}$$

6v



$$Q = -5.23 \text{ MeV}$$

$$K_p^{\text{in}} = 5.26 \text{ MeV.}$$

$$\text{but } E_c = 12.33 \text{ MeV}$$

} ?